

GENKWANIN AND IRIDOID GLYCOSIDES FROM *Leonurus turkestanicus*

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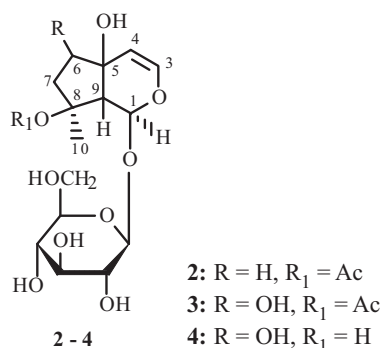
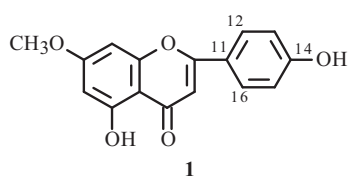
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Leonurus cardiaca L. (Labiatae) and *L. quinquelobatus* Gilib. are used in galenics as sedative and hypotensive agents [1]. Two species of *Leonurus* are indigenous to Uzbekistan, *L. turkestanicus* V. Krecz. et Kupr. and *L. pancerooides* M. Pop. [2], and are used as medicinal plants instead of the official species. However, the chemical compositions of these plants have not been reported. This prompted us to study *L. turkestanicus*.

The air-dried aerial part of *L. turkestanicus* (2 kg) that was collected in Kara-Tepe, Tashkent Oblast on June 30, 2007, during flowering was extracted with MeOH (4 × 14 L). The MeOH extract was evaporated to a thick condensate and dissolved in MeOH (70%, 1 L). The solution was filtered. The filtrate was evaporated to remove completely MeOH. The solution volume was adjusted to 0.5 L with H₂O. The aqueous solution was extracted three times with *n*-BuOH. The *n*-BuOH extract was evaporated to dryness and dried. The yield of the *n*-BuOH extract was 63.42 g. The *n*-BuOH extract was chromatographed over a column of KSK silica gel with elution successively by CHCl₃, CHCl₃:MeOH (1, 100:1; 2, 50:1; 3, 10:1), and CHCl₃:MeOH:H₂O (4, 70:12:1).

Fractions eluted by CHCl₃ and system 1 contained sterols.

System 2 eluted flavonoid **1** (40 mg).



Further elution of the column by system 3 isolated a fraction containing **2**, rechromatography of which by the same system afforded **2** (6 mg).

Then, elution of the column by system 4 isolated **3** and **4**. A part of the fractions was rechromatographed over the column using the same system in order to purify the compounds from a yellow pigment. This afforded **3** (1.3 g) and **4** (0.756 g), which were the principal constituents of the extracted compounds. The aqueous phase remaining after work up with *n*-BuOH contained exclusively **3** and **4** in about a 1:1 ratio. Evaporation of the H₂O afforded these two compounds (120 g). This led to the conclusion that the content of **3** and **4** in the air-dried plant was at least 6%.

Genkwanin (1), C₁₆H₁₂O₅. Inspection of the PMR and ¹³C NMR spectra of **1** (Table 1) showed that it was a 5,7,14-trisubstituted flavonoid, two substituents of which were hydroxyls. The third substituent was a methoxyl. The difference nuclear Overhauser effect (NOE) with saturation of the methoxyl exhibited couplings for the H-6 and H-8 resonances. This determined unambiguously the location of the methoxyl on C-7. Therefore, **1** had the structure 5,14-dihydroxy-7-methoxyflavone. Genkwanin has the same structure [3, 4].

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Table 1. Chemical Shifts of C and H Atoms in **1**

C atom	DEPT	Data in C ₅ D ₅ N		Data in (CD ₃) ₂ CO	
		δ _C	δ _H (J/Hz)	δ _C	δ _H (J/Hz)
2	C	164.83	–	164.01	–
3	CH	103.95	6.82 s	102.96	6.77 s
4	C	182.80	–	181.82	–
5	C	158.07	–	157.16	OH 12.90 s
6	CH	98.57	6.50 d (2.4)	97.86	6.31 d (2.2)
7	C	165.80	–	165.05	–
8	CH	92.83	6.58 d (2.4)	92.60	6.69 d (2.2)
9	C	162.68*	–	161.18	–
10	C	105.87	–	104.61	–
11	C	121.96	–	121.01	–
12	CH	128.98	7.83 d (8.9)	128.45	7.89 d (8.8)
13	CH	116.95	7.15 d (8.9)	115.89	6.88 d (9)
14	C	163.05*	–	161.18	–
15	CH	116.95	7.15 d (8.9)	115.89	6.88 d (9)
16	CH	128.98	7.83 d (8.9)	128.45	7.89 d (8.8)
CH ₃ O	CH ₃	55.90	3.65 s	55.95	3.81 s

*Assignment of resonances is arbitrary.

Table 2. Chemical Shifts of C and H Atoms in **2-4**

C atom	DEPT	2 (C ₅ D ₅ N)		3 (C ₅ D ₅ N)		3 (D ₂ O)		4 (D ₂ O)	
		δ _C	δ _H (J/Hz)	δ _C	δ _H (J/Hz)	δ _C	δ _H (J/Hz)	δ _C	δ _H (J/Hz)
1	CH	94.09	6.40 d (1.6)	94.35	6.57 d (1.5)	93.97	6.06 d (1.4)	93.38	5.74 d (1.4)
3	CH	141.07	6.39 d (6.4)	141.81	6.39 d (6)	142.81	6.46 d (6.4)	141.83	6.39 d (6.4)
4	CH	110.27	5.04 dd (6.3, 1.6)	107.76	5.01 dd (6.4, 1.7)	104.93	5.02 dd (6.5, 1.7)	107.02	5.07 dd (6.4, 1.6)
5	C	72.10	–	72.82	–	72.48	–	71.75	–
6	CH(CH ₂)	37.14	1.63 ddd (11.9, 6.5, 1.5), 2.21 m	76.48	3.96 m	76.42	3.84 dd (4.4, 1.5)	76.92	3.83 t (4.7, 4.7)
7	CH ₂	38.23	1.53 m, 1.96 ddt (13.6, 6.8, 1.3, 1.3)	45.43	2.41 dt (15, 1.2, 1.2), 1.93 dd (14.9, 4.7)	44.45	2.03 dd (15.6, 4.6), 2.18 dt (15.7, 1.2, 1.2)	46.01	1.84 ddd (14.2, 4.2, 1), 2.02 dd (14.1, 5)
8	C	88.27	–	86.68	–	88.06	–	77.65	–
9	CH	58.06	3.04 q (1.5)	54.84	3.41 q (1.5)	53.17	2.87 q (1.4)	57.51	2.57 q (1.2)
10	CH ₃	21.89	1.42 s	22.26	1.46 s	21.41	1.45 s	24.59	1.27 s
Ac	CH ₃	21.68	1.72 s	21.70	1.71 s	21.81	2.06 s	–	–
	–	170.88	–	170.53	–	174.30	–	–	–
<i>β</i> -D-Glcp									
1	CH	98.64	5.13 d (8)	98.75	5.18 d (8)	98.82	4.74 d (8)	98.92	4.74 d (8)
2	CH	74.71	3.88 dd (9, 8)	74.46	3.93 dd (9, 8)	72.62	3.32 dd (9, 8)	73.09	3.33 dd (9.4, 8)
3	CH	78.45	4.11 m	78.11	4.15 t (8.8)	75.56	3.52 t (9)	75.96	3.52 t (9)
4	CH	71.59	4.11 m	71.25	4.09 t (8.8)	69.74	3.42 dd (9.7, 9)	70.26	3.41 dd (9.8, 9)
5	CH	78.68	3.86 m	78.42	3.86 ddd (9, 5.4, 2.2)	76.29	3.49 m	76.86	3.51 m
6	CH ₂	62.63	4.21 dd (11.9, 5.4), 4.37 dd (11.9, 2.5)	62.36	4.18 dd (12, 5.6), 4.39 dd (12, 2.4)	60.79	3.75 dd (12.5, 5.7), 3.94 dd (12.5, 2.2)	61.29	3.74 dd (12.3, 5.8), 3.93 dd (12.4, 2.3)

*Assignment of resonances is arbitrary. H-9 has one constant ³J and two ⁴J of similar value. Because of this the H-9 resonance is observed as a 1:3:3:1 quartet.

8-Acetylharpagide (3), C₁₇H₂₆O₁₁, and **harpagide (4)**, C₁₅H₂₄O₁₀. Interpretation of PMR and ¹³C NMR spectra of **3** (Table 2) and DEPT, COSY, Hetcor, and NOESY spectra were consistent with a structure identical with that of 8-acetylharpagide [5]. Analogously, **4** was identified as harpagide [6]. As expected, alkaline hydrolysis of 8-acetylharpagide gave a product that was identified as **4**.

6-Deoxy-8-acetylharpagide (2), C₁₇H₂₆O₁₀. PMR and ¹³C NMR spectra of **2** showed that the compound differed from 8-acetylharpagide by the lack of a secondary hydroxyl on C-6, i.e., the compound was 6-deoxy-8-acetylharpagide. The iridoid glycoside was identical to that reported [7].

Thus, genkwanin (**1**), 6-deoxy-8-acetylharpagide (**2**), 8-acetylharpagide (**3**), and harpagide (**4**) were isolated and identified from the aerial part of *L. turkestanicus*. It is quite certain that the sedative and hypotensive properties of *L. turkestanicus* are due to iridoid glycosides.

PMR spectra were recorded from C₅D₅N and DMSO-d₆ (δ, ppm, 0 = HMDS) and D₂O solutions on a UNITYplus 400 (Varian) spectrometer. The internal standard for the last solution was acetone, the resonance of which had chemical shift δ 2.22 or *tert*-butyl alcohol, the resonance of which was found at δ 1.24. ¹³C NMR spectra were also recorded on the UNITYplus 400 spectrometer (Varian) with full suppression of C–H coupling and under DEPT conditions. 2D NMR spectra (¹H–¹H COSY, Hetcor, NOESY) were obtained using standard Varian programs. Chemical shifts of C atoms are given relative to the β-carbon atoms of C₅D₅N (δ 123.493 vs. TMS). Spectra taken from D₂O solutions have chemical shifts vs. that of CD₃OD (δ 49.00).

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